

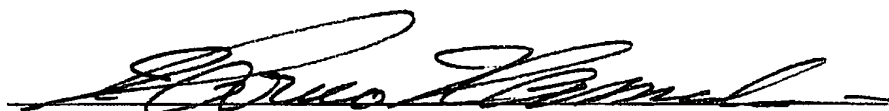
## VERIFICATION OF THE TRANSLATION

I, the below-named Chartered Patent Attorney of Tokyo Japan having an office at an address stated below, hereby declare that :

I am knowledgeable in the English and Japanese languages, and I believe that the attached English translation of the Japanese Patent Application No. 2002-326723 filed on November 11, 2002 is a true and complete translation of said application.

I also hereby declare that all statements made herein of my own knowledge are true and that all statements made on information and belief are believed to be true ; and further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under Section 1001 of Title 18 of the United States Code, and that such willful false statements may jeopardize the validity of the application or any patent issuing thereon.

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Patent Attorney Haruo HAMADA  
4-12, MINAMI-AOYAMA 3-CHOME,  
MINATO-KU, TOKYO JAPAN



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[Addressee] The Commissioner of the Patent Office

[Inventor]

[Address] c/o NIPPON CHEMI-CON CORP  
167-1, Higashi-Ome 1-chome, Ome-shi,  
Tokyo, Japan

[Name] Minoru Wada

[Inventor]

[Address] c/o Mitsubishi Chemical, Inc.  
3-1, Chuo 8-chome, Amimachi, Inashiki-gun,  
Ibaraki-ken, Japan

[Name] Masayuki Takeda

[Inventor]

[Address] c/o Mitsubishi Chemical, Inc.  
3-1, Chuo 8-chome, Amimachi, Inashiki-gun,  
Ibaraki-ken, Japan

[Name] Makoto Ue

[Applicant]

[Identification Number] 000228578

[Name] NIPPON CHEMI-CON CORP.

[Representative] Hikokichi Tokiwa

[Applicant]

[Identification Number] 000005968

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[Name] Mitsubishi Chemical, Inc.

[Representative] Ryuichi Tomizawa

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[Kind of Document] SPECIFICATION

[Title of the present Invention]

ELECTROLYTE SOLUTION FOR ELECTROLYTIC CAPACITOR  
AND THE ELECTROLYTIC CAPACITOR USING IT

[CLAIMS]

[Claim 1] An electrolyte solution for an electrolytic capacitor containing an aluminum tetrafluoride salt and a solvent with high boiling point.

[Claim 2] An electrolyte solution for an electrolytic capacitor according to claim 1 wherein the solvent with high boiling point is sulfolane, 3-methyl sulfolane, or 2,4-dimethyl sulfolane.

[Claim 3] An electrolytic capacitor using the electrolyte solution of claims 1 or 2.

[Detailed Description of the Invention]

[0001]

[Industrial Application]

The present invention relates to an electrolytic capacitor for an electrolytic capacitor and the electrolytic capacitor using it, especially, the electrolyte solution and the electrolytic capacitor using it having a low

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impedance characteristic and a high withstand voltage characteristic.

[0002]

[Description of the Prior Art]

An electrolytic capacitor typically has an anode electrode foil made of a band-shaped high purity aluminum foil where the effective aluminum foil surface has been enlarged through etching process chemically or electrochemically, and an oxide film is formed on the surface, through a chemical process of treating the aluminum foil with a chemical solution such as ammonium borate aqueous solution and the like. A cathode electrode foil is also made of an etched aluminum foil of high purity. Capacitor element is formed by the anode electrode foil and the cathode electrode foil, wound together with intervening separator made of manila paper and the like. Next, the capacitor element, after impregnating with an electrolyte solution for driving the electrolytic capacitors, is housed into a bottomed outer case made of aluminum and the like. The outer case is equipped at the opening with a sealing member made of an elastic rubber, and is sealed by drawing.

[0003]

Herewith, as electrolyte solution for driving the electrolytic capacitor having high conductivity, and to be impregnated to the capacitor element, wherein  $\gamma$ -butyrolactone is employed as the main solvent composed of quaternized cyclic amidin compounds (imidazolinium cation and imidazolium cation) as the cationic component and acid conjugated

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bases as the anionic component are dissolved therein as the solute (refer to the patent documents 1 and patent documents 2).

[0004]

[Patent documents 1] JP-H08-321440-A

[Patent documents 2] JP-H08-321441-A

[0005]

[Problem(s) to be Solved by the Invention]

However, due to the remarkable development of digital information devices in recent years, the high-speed driving frequency of micro-processor which is a heart of these electronic information devices is in progress. Accompanied by the increase in the power consumption of electronic components in the peripheral circuits, the ripple current is increased abnormally, and there is a strong demand for the electrolytic capacitors used in these circuits to have a low impedance characteristic.

[0006]

Moreover, in the field of vehicles, with the recent tendency toward improved automobile functions, a low impedance characteristic is in high demand. By the way, the driving voltage of the vehicle circuit of 14V has been progressed to 42V accompanied by the increase in the power consumption. To comply with such a driving voltage, the electrolytic capacitor requires the withstand voltage characteristic of 28V and 84V and

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more. Furthermore, in this field, there is a demand of using high temperature, an electrolytic capacitor, high temperature life characteristic is required.

[0007]

However, in the aforementioned the electrolyte solution for the electrolytic capacitor, it could not respond to such a low impedance characteristic, and its limit withstand voltage was 30V. The electrolytic capacitor was able to respond to 28V but could not reply to the demand of high withstand voltage, such as 84V and more.

[0008]

The present invention aims to supply electrolyte solution for an electrolytic capacitor and the electrolytic capacitor using it, having an excellent high temperature characteristic, a low impedance characteristic, and a high withstand voltage characteristic of 100V class.

[0009]

[Means for Solving the Problem]

The present invention is characterized in that electrolyte solution for the electrolytic capacitor contains an aluminum tetrafluoride salt and a solvent with high boiling point.

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Moreover, the present invention is characterized in that the sulfolane, 3-methyl sulfolane, and 2,4-dimethyl sulfolane are used as the solvent with high boiling point.

[0011]

Further, the electrolytic capacitor of the present invention utilizes the aforementioned electrolyte solution.

[0012]

[Embodiment of the Invention]

The electrolyte solution for the electrolytic capacitor used in the present invention contains an aluminum tetrafluoride salt and a solvent with high boiling point, and a high sparking voltage, and a low specific resistance characteristic. The high temperature stability is also excellent.

[0013]

As the aluminum tetrafluoride salt constituting the aluminum tetrafluoride as anion component, examples of this salt include an ammonium salt, an amine salt, a quaternary ammonium salt, or a quaternary cyclic amidinium ion as cation component, can be used. Examples of an amine constituting the amine salt include a primary amine (such as methylamine, ethylamine, propylamine, butylamine, ethylenediamine, monoethanolamine, and the like); secondary amine (such as dimethylamine, diethylamine, dipropylamine, ethy-methylamine,



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diphenylamine, diethanolamine and the like); and tertiary amine (such as trimethylamine, triethylamine, tributylamine, 1,8-diazabicyclo[5,4,0]undecen-7, triethanolamine, and the like). Examples of a quaternary ammonium constituting the quaternary ammonium salt include a tetraalkylammonium (such as tetramethylammonium, tetraethylammonium, tetrapropylammonium, tetrabutylammonium, methyltriethylammonium, di-methyldiethylammonium and the like) and a pyridinium (such as 1-methylpyridinium, 1-ethylpyridinium, 1,3-diethylpyridinium and the like).

[0014]

Furthermore, as for salt containing the quaternized cyclic amidinium ion as a cationic component, the quaternized cyclic amidinium ion is a cation formed by quaternized a cyclic compound having an N,N,N'-substituted amidine group, and the following compounds are exemplified as the cyclic compound having an N,N,N'-substituted amidine group. They are an imidazole monocyclic compound (for example, an imidazole homologue, such as 1-methylimidazole, 1-phenylimidazole, 1,2-dimethyl-imidazole, 1-ethyl-2-methylimidazole, 2-ethyl-1-methylimidazole, 1,2-diethylimidazole, 1,2,4-trimethylimidazole and the like, an oxyalkyl derivative, such as 1-methyl-2-oxymethylimidazole, 1-methyl-2-oxyethyl-imidazole, and the like, a nitro derivative such as 1-methyl-4(5)-nitroimidazole, and the like, and an amino derivative such as 1,2-dimethyl-5(4)-aminoimidazole, and

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the like), a benzoimidazole compound (such as 1-methylbenzoimidazole, 1-methyl-2-benzylbenzoimidazole, 1-methyl-5(6)-nitrobenzo-imidazole and the like), a compound having a 2-imidazoline ring (such as 1-methylimidazoline, 1,2-dimethylimidazoline, 1,2,4-trimethylimidazoline, 1-methyl-2-phenylimidazoline, 1-ethyl-2-methylimidazoline, 1,4-dimethyl-2-ethyl-imidazoline, 1-methyl-2-ethoxymethylimidazoline, and the like), a compound having a tetrahydropyrimidine ring (such as 1-methyl-1,4,5,6-tetrahydropyrimidine, 1,2-dimethyl-1,4,5,6-tetrahydropyrimidine, 1,5-diazabicyclo[4,3,0]-nonene-5, and the like), and the like.

[0015]

Examples of the solvent with high boiling point utilized in the present invention include the followings. Namely, sulfolane series such as sulfolane, 3-methyl sulfolane, 2,4-dimethyl sulfolane, and the like, 1,3-diethyl-2-imidazolidinone, 1,3-dipropyl-2-imidazolidinone, 1-methyl-3-ethyl-2-imidazolidinone, and the like; 1,3,4-trialkyl-2-imidazolidinone such as 1,3,4-trimethyl-2-imidazolidinone, and the like; 1,3,4,5-tetraalkyl-2-imidazolidinone such as 1,3,4,5-tetramethyl-2-imidazolidinone, and the like, 3-ethyl-1,3-oxazolidinone-2-one, and the like; and carbonates such as ethylene carbonate, propylene carbonate, and the like. Among these, sulfolane, 3-methyl sulfolane, and 2,4-dimethyl sulfolane are preferably used because of their excellent moisture characteristics. That is,

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normally, a cathode current passes through the cathode electrode lead to generate hydroxyl ion, and the basicity increases. However, these solvents do not decompose even at pH value of 7 and more, which is effective in preventing the leakage caused by decomposition of solvents in case of using the quaternary ammonium salt and the quaternary cyclic amidium salt.

[0016]

The solvent in use for an electrolyte solution according to the present invention comprises a polar protic solvent, an aprotic polar solvent, and their mixture thereof can be used. Examples of the polar protic solvent include monohydric alcohols (such as ethanol, propanol, butanol, pentanol, hexanol, cyclo-butanol, cyclo-pentanol, cyclo-hexanol, benzyl alcohol, and the like); and polyhydric alcohol and oxy alcohol compounds (such as ethylene glycol, propylene glycol, glycerine, methyl cellosolve, ethyle cellosolve, methoxy propylene glycol, dimethoxy propanol, and the like). Moreover, representative examples of the aprotic polar solvent include amide series (such as N-methylformamide, N,N-dimethylformamide, N-ethylformamide, N,N-diethylformamide, N-methyl acetamide, N,N-dimethyl acetamide, N-ethyl acetamide, N,N-diethyl acetamide, hexamethylphosphoric amide, and the like); lactone compounds (such as  $\gamma$ -butyrolactone,  $\delta$ -valerolactone,  $\gamma$ -valerolactone, and the like); cyclic amide compounds (such as N-methyl-2-pyrrolidone, and the like); nitrile compound (such as acetonitrile, and the like); sulfoxide

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compound (such as dimethyl sulfoxide, and the like); and the like.

[0017]

Further, the content of solvents with high boiling point in the solvent is less than 25% by weight, preferably is ranging from 1 to 20% by weight, and more preferably is ranging from 3 to 15% by weight. The specific resistance increases beyond this range, and the high temperature stability increases beyond the lower limit.

[0018]

According to electrolyte solution for an electrolytic capacitor of the present invention has the low impedance characteristic, the high sparking voltage, and the excellent high temperature stability. The electrolytic capacitor using this electrolyte solution has a low impedance characteristic and high withstand voltage characteristic of 100V class, and an excellent high temperature life characteristic.

[0019]

[Embodiments]

Subsequently, the present invention will be explained by using the embodiments. A capacitor element is formed by winding an anode electrode foil and a cathode electrode foil via a separator. The anode electrode foil and the cathode electrode foil are connected respectively to a lead wire for leading the anode electrode and an another lead wire for

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leading the cathode electrode.

[0020]

These lead wires are composed of connecting members being in contact with the electrode foils, and the rod members having been molded integrally with the connecting members, and outer connecting members having been fixed at the tip of the rod members. The connecting member and the rod member are made from aluminum of 99% purity while the outer connecting member is made of a copper-plated steel wire (hereinafter CP wire). On the surfaces of the rod members of the lead wires at least, anode oxide films made of aluminum oxide are formed by a chemical treatment with ammonium phosphate aqueous solution. These lead wires are connected respectively to the electrode foils at the connecting members by means of stitching, ultrasonic welding, and the like.

[0021]

The anode electrode foil is made of an aluminum foil of 99.9% purity in an acidic solution thereby enlarging the surface area thereof through the chemical or electrochemical etching process, and then subjecting the aluminum foil to a chemical treatment in an ammonium adipate aqueous solution, to thereby form an anode oxide film on the surface thereof.

[0022]

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The capacitor element, which impregnates the electrolyte solution, is then housed into a bottomed outer case made of aluminum. The outer case is provided at the opening with a sealing member and then sealed by drawing. The sealing member has perforation holes through which the lead wires are to be passed.

[0023]

(Table 1) shows the electrolyte solution for electrolytic capacitor used in the present invention. As comparative example, the conventional electrolyte solution containing 90% by weight of  $\gamma$ -butyrolactone, and 10% by weight of hydrogen phthalate 1-ethyl-2, 3-dimethyl-imidazolinium is used. The sparking voltage is 105V, and the specific resistance is 142  $\Omega$  cm.

[0024]

According to the electrolytic capacitors which were constructed by using the electrolyte solution of above embodiments, the rated values of the electrolytic capacitors were 100WV-33 $\mu$ F, and the characteristics of the electrolytic capacitor were evaluated. The test conditions are 125°C and 1,000 hours in the loaded state. The results are shown in (Table 2).

[0025]

[Table 1]

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	GBL	SL	A	Sparking voltage (V)	Specific resistance ( $\Omega$ c m)
Embodiment 1	85	5	10	210	77
Embodiment 2	80	10	10	207	80
Embodiment 3	75	15	10	211	84
Embodiment 4	70	20	10	204	88
Comparative example	90	—	10	205	73

GBL :  $\gamma$ - butyrolactone

SL : sulfolane

A : Tetraaluminate 1-ethyl-2,3-dimethyl-imidazolinium

[0026]

[Table 2]

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	Initial Characteristic		125°C - 1000 hours	
	Cap ( $\mu$ F)	tan $\delta$	$\Delta$ Cap (%)	tan $\delta$
Embodiment 1	33.6	0.014	-2.2	0.019
Embodiment 2	33.5	0.014	-2.2	0.019
Embodiment 3	33.6	0.015	-2.1	0.018
Embodiment 4	33.5	0.015	-2.0	0.017
Comparative example	33.5	0.014	-2.0	0.023

Cap: electrostatic capacity

tan $\delta$ : tangent of dielectric loss $\Delta$ Cap: change in electrostatic capacity

[0027]

As (Table 1) clearly shows, the sparking voltage of the electrolyte solution for the electrolytic capacitor of the embodiment is high, and the specific resistance ratio is low, compared with that of the conventional example. Also, as can be seen from (Table 2), the dielectric loss of coefficient (tan  $\delta$ ) of the electrolytic capacitor of 100WV using this is low, and the high temperature life characteristic is excellent compared with that of the comparative example. The effectiveness of the present invention is apparent by replacing the sulfolane with the 1, 3-diethyl-imidazolidinone.



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[0028]

[Effect of the Invention]

According to the present invention, the electrolyte solution for electrolytic capacitor containing the aluminum tetrafluoride salt and the solvent with high boiling point is used, such that the electrolytic capacitor having a low impedance characteristic and high withstand voltage characteristic, and excellent high temperature life characteristic are provided.